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SYNTHESIS AND ANTIMICROBIAL STUDIES OF INDOLYL PYRIMIDINES

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Abstract : Many classes of chemotherapeutic agents containing pyrimidine nucleus are in clinical use such as antibacterial (Sulfadiazine,sulfamerazine and sulfamethazine), anticancer (5- fluorouracil and ftorafur), antiviral (iodoxuridine, trifluridine and zidovudine)agents. The reaction of indolyl chalcone with urea or thiourea gave indolyl pyrimidines derivatives. All the synthesized compounds have been characterized by elemental and spectral (IR, PMR and Mass) analyses. All representative compounds have been evaluated for their antibacterial and antifungal activities.

Keyword: Indolyl pyrimidines, thiourea, indoles

Introduction

The indole derivatives are known to posses anticancer, antioxidant, antirheumatoidal and anti HIV activities. indolyl pyrimidines systems as antioxidants, DNA cleavage and cytotoxic agents. In continuation if our interest on drug like molecules. The evaluation of the novel indolyl pyrimidine analogues for antimicrobial activities to generate novel molecular templates which are likely to exhibit interesting biological properties.

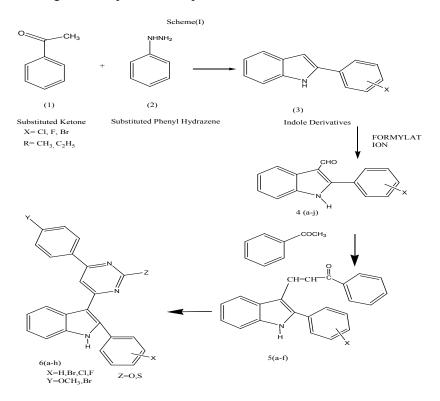
Pyrimidine present an interesting group of compounds many of which possess widespread pharmacologic properties such as antimicrobial ¹, analgesic, antiviral, anti-inflammatory ², anti HIV³, antitubercular⁴, antitumor⁵, antineoplastic⁶, antimalerial⁷, diuretic⁸, cardiovascular ⁹ agents, hypnotic drugs for the nervous system ¹⁰, calcium sensing receptor antagonists ¹¹ and also for antagonists of the human A_{2A} adenosine receptor¹². Several drugs have been developed as anticancer agents which contain pyrimidine moieties, such as clofarabine, capeitabine, cytarabine, fludarabine, gemcitabine, decitabine ¹³ and floxuridine ^{14,15}.

Pyrimidine based heterocycles are potential bioactive molecules and exhibit antibacterial, anti-inflammatory, cytotoxic¹⁶, antitumoral, analgesic, antitubercular and antiviral, antihypertensive, anticonvulsant, antimicrobial agents and also act as enzyme inhibitors. Many classes of chemotherapeutic agents containing pyrimidine nucleus are in clinical use such as antibacterial (Sulfadiazine,sulfamerazine and sulfamethazine), anticancer (5- fluorouracil and ftorafur), antiviral (iodoxuridine, trifluridine and zidovudine), antifungal

(flucytocine) and antimalarial (pyrimethamine)agents. Nitrogen containing heterocycles such as pyrimidine and indole is a promising structural moiety for drug designing.

Experimental

Substituted 2-phenyl indoles were prepared by Fischer-indole synthesis and by the method of Joshi et al. Substituted 2-phenyl indoles was subjected to Vilsmeier – Haack formylation ^a with POCl₃ and N, N- dimethylformamide to give 2-arylindole-3- carboxaldehyde (4) was reacted with substituted acetophenones in ethanolic NaOH to obtain chalcone (5 a-f), which were condensed with urea and thiourea in presence of cyclising agent, concentrated hydrochloric acid to obtain substituted pyrimidines (6a-h) respectively. The title compound gave a single spot on TLC in different solvent system. The synthetic sequence leading to the formation of targeted compounds is depicted in Scheme – I



Materials and Methods

Characterization of synthesized compounds has been done on the basis of elemental analyses, IR and ¹H NMR studies. C and H analyses of compounds has been done using coleman C and H analyzer .Nitrogen analyses has been done using coleman N-analyses 29. Melting point were determined in open glass capillaries and are uncorrected. IR (4000-400 cm ⁻¹) were recorded on Perkin Elmer model 557 and Nicholet magna model 750 spectrophotometer in KBr pallets at central Drug Research Institute (CDRI), Lucknow .¹H NMR spectra were recorded on spectrometer (300 mHz) using CDCl ₃ /DMSO as solvent.TMS was taken as standard. The chemical shift are in δ ppm .The purity of compounds was checked by TLC using silica gel-G as adsorbent in various solvent system. Visualization was accomplished by U.V light or iodine adsorption.

M. Mangal et al. / Heterocyclic Letters Vol. 6| No.4|795-803|Aug-Oct| 2016

1- (4- substituted phenyl) -3-(2'- aryl indolyl) -2- propen-1-one (5a-f)

Equimolar quantities (0.01 M) of 2-arylindole-3- carboxaldehyde and substituted acetophenones were taken in 100 ml conical flask and dissolved in 20ml of ethanol to this (.03mol) of NaOH in minimum quantity of water was added . The mixture was stirred on magnetic stirrer and the reaction was monitored with TLC . Reaction mixture was diluted with water and acidified with concentrated hydrochloric acid . The precipitated chalcon was filtered and recrystallized from absolute ethanol. The purity of the chalcones was tested with thin layer chromatography using solvent system (50% benzene and 50% petroleum ether). Compounds are white and pale yellow. The physical and analytical characteristics are given in **Table-1**

4- (2-arylindol-3- yl)-6-(4- substituted phenyl) -2- substituted pyrimidines (6a-h)

Chalcone (0.01mol) and urea or thiourea (.01mol) were dissolved in absolute alcohol (20ml) few drops of concentrated HCl were added and the reaction mixture was refluxed and the reaction was monitored by TLC. After completion of reaction it was poured into 250ml of ice cold water and kept for some time. The crude solid was filtered and subjected to column chromatography. Elution with solvent system ethyl acetate / petroleum ether (60-80°). The compounds were blue in colour. The physical and analytical characteristics are given in **Table-2**

Result and Discussion

Substituted 2-phenyl indoles were prepared by Fischer indole synthesis and by the method of Joshi et al. (Scheme-I). The IR spectra of 2-arylindole showed absorption band at 3450-3350 cm^{-1} which is attributed to >N-H stretching vibration. In the IR spectra of the compounds (4a-j) >N-H absorption band appears at 3417-3400 cm⁻¹ and aromatic C-H str. Absorption peak appears at 3060-3051 cm⁻¹. Aliphatic C-H stretching Vibration is observed at 2839 cm⁻¹ aromatic C=C absorption band is observed at 1604 cm⁻¹ and HC=O absorption appears at 1750 cm⁻¹ .Absorption band due to C-Br appears at 548 cm⁻¹.The ¹H NMR spectra of 2arylindole revealed the presence of a broad resonance signal in the region of δ 7.8-8.0ppm which is attributed to >N-H proton, a methine resonance signal (=C-H, at C-3) appears as a sharp singlet at δ 6.4ppm. In the ¹H NMR spectra of all compounds (4a-j), the disappearance of resonance signal from δ 6.4ppm due to methine =C-H proton at C-3 position of indole moiety, supports the formation of formyl indole. In the IR spectra of (5a-f) shows the disappearance of >C=O absorption band at 1750 cm⁻¹ supports the formation of 1- (4substituted phenyl) -3-(2'- aryl indolyl) -2- propen-1-one. The IR and ¹HNMR spectra of the title compounds are given in Table-3.

Antibacterial and Antifungal Activities

All the synthesized compounds were screened for their antimicrobial activity against bacteria *Escherichia coli* and *Bacillus subtilis* and fungi *Candida albicans*, *fuserium oxysporium* at different concentration by disc diffusion method. Streptomycin and Ketokenazole are used as standard for evaluating antibacterial and antifungal activities respectively. Some compounds show prominent results. The antibacterial and antifungal activities are listed in **Table-4 & 5**.

Table – 1	
Physical characteristics and analytical Characteristics of 1	- (4- substituted phenyl) -3-
(2'- aryl indolyl) -2- propen-1-one (5a-f)	

Compd.No.	x	Y aniline	M.P.(°C)	M.F.	C (%)		H (%)		N (%)	
					Cal.	Obs.	Cal.	Obs.	Cal.	Obs.
5a	Н	Br	180	C ₂₃ H ₁₆ Br N O	68.65	68.62	3.98	3.93	3.48	3.43
5b	Br	Br	250	C ₂₃ H ₁₅ Br ₂ N O	57.38	57.34	3.11	3.10	2.91	2.90
5c	Cl	Br	270	C ₂₃ H ₁₅ Br Cl N O	63.23	63.21	3.43	3.40	3.20	3.19
5d	Н	OCH ₃	258	$\begin{array}{cc} C_{24}H_{19} & N \\ O_2 \end{array}$	81.58	81.53	5.38	5.34	3.96	3.92
5e	Br	OCH ₃	258	C ₂₄ H ₁₈ Br N O ₂	66.66	66.62	4.16	4.14	3.24	3.21
5f	F	Br	268	C ₂₃ H ₁₅ Br FN O	65.71	65.70	3.57	3.53	3.33	3.31

Table –2

Physical characteristics and analytical Characteristics of 4- (2-arylindol-3- yl)-6-(4-substituted phenyl) -2- substituted pyrimidines (6a-h)

Compd.No .	x	Y anilin	Z	M.P.(°C	ме	С	(%)	Н	(%)	Ν	(%)		
		e	e) M.)	M.F.	Cal.	Obs.	Cal	Obs ·	Cal.	Obs.
6a	B r	Br	0	180	$\begin{array}{c} C_{24}H_1\\ {}^5\\Br_2N_3\\O\end{array}$	55.2 7	55.2 1	2.8 7	2.82	8.06	8.02		
6b	Cl	Br	0	182	C ₂₄ H ₁ 5 Br Cl N ₃ O	60.4 4	60.4 2	3.1 4	3.11	8.81	8.80		

6c	Н	OCH ₃	0	184	$\begin{array}{c} C_{25}H_1\\ {}_9 \qquad N_3\\ O_2 \end{array}$	76.3 3	76.3 0	4.8 3	4.80	10.6 8	10.6 3
6d	Н	Br	0	240	C ₂₄ H ₁ ₆ Br N ₃ O	65.1 5	65.1 3	3.6 1	3.60	9.50	9.49
6e	B r	OCH ₃	0	260	C ₂₅ H ₁ ₈ Br N ₃ O ₂	63.5 5	63.5 1	3.8 1	3.80	8.89	8.84
6f	F	Br	0	265	$\begin{array}{c} C_{24}H_1\\ {}_5 Br\\ F N_3\\ O \end{array}$	62.6 0	62.5 9	3.2 6	3.23	9.13	9.10
6g	B r	Br	S	265	C ₂₅ H ₁ 5 Br ₂ N ₃ S	54.6 4	54.6 2	2.7 3	2.70	7.65	7.61
6h	Cl	Br	S	250	$\begin{array}{cc} C_{25}H_1\\ {}_5 & Br\\ Cl & N_3\\ S \end{array}$	59.4 6	59.4 2	2.9 7	2.92	8.32	8.30

M. Mangal et al. / Heterocyclic Letters Vol. 6| No.4|795-803|Aug-Oct| 2016

Table- 3

Spectral	data	of 4- (2-arylindol-3-	yl)-6-(4-	substituted	phenyl) -	2- substituted
pyrimidine	s (6a-h)						

Compd. No	IR(KBr) v _{max} cm ⁻¹	¹ HNMR(CDCl ₃) δ ppm	FEB mass m/z
6a	3403 (NH str.), 3050 (Aromatic C-H Str.), 2839 (Aliphatic C-H Str.), 1750 (C=O), 1657(aromatic C=C Str.), 1601(C=C Str.).	11.2 {s,1H,OH),10.9(s,1H,NH,), 8.3-7(m,5H,Ar-H),7.6(s,1H Pyrimidine proton), 7.3(s,1H,Indolylproton),7.1- 6.7(M,8H,Ar-H),	521/523/525 isotopic cluster
6c	3416 (NH str.), 3049 (Aromatic C-H), 2839 (Aliphatic C-H Str.), 1716 (C=O), 1657(aromatic C=C Str.), 1602(C=C Str.).	11.3(s,1H, OH),10.8(S,1H,NH.), 8.4-7.7(m,4H,Ar-H), 7.5(S,1H,Pyrimidine proton), 7.3(S,1H,Indolyl proton), 7.1-6.7(M,8H,Ar-H), 5.2(S,2H,NH),	393 (Molecular ion)
6g	3323 (NH str.), 2961 (Aromatic C- H), 2826 (Aliphatic C-H Str.), 1750 (C=O), 1598 (aromatic C=C Str.), 748(C-Cl Str.), 499 (C-Br Str.).	10.7(S,1H,NH,Exchangeable), 8.4(S,1H,Pyrimidine proton), 8.3-7.4(M,4H,Ar-H), 7.3(S,1H,Indolyl proton), 7.0-6.6(m,8HAr-H),	549/551/ 553 isotopic cluster

Table- 4Antibacterial activityof 4- (2-arylindol-3- yl)-6-(4- substituted phenyl) -2- substitutedpyrimidines (6a-h)

compounds	Mean valu	ie of area of	Mean valu	e of area of	Mean value of area of		
	inhibition	in mm	inhibition	in mm	inhibition	in mm	
	(400ppm)		(800ppm)		(1000ppm)		
	E.coli	Bacillus	E.coli	Bacillus	E.coli	Bacillus	
		subtilis		subtilis		subtilis	
Streptomycin	3	6	5	7	10	12	
6a	NIL	NIL	NIL	20mm	8mm	6mm	
6b	NIL	NIL	NIL	NIL	4mm	NIL	
6c	6mm	14mm	10mm	NIL	10mm	NIL	
6d	6mm	NIL	6mm	NIL	12mm	16mm	
6e	10mm	NIL	NIL	NIL	NIL	NIL	
6f	NIL	NIL	NIL	NIL	10mm	NIL	
6g	NIL	NIL	NIL	NIL	NIL	NIL	
6h	NIL	NIL	NIL	NIL	NIL	8mm	

Table- 5 Antifungal activity of 4- (2-arylindol-3- yl-)-6-(4- substituted phenyl) -2- substituted pyrimidines (6a-h)

Compounds	Mean valu	ie of area of	Mean valu	ie of area of	Mean value of area of		
	inhibition	in mm	inhibition	in mm	inhibition	in mm	
	(400ppm)		(800ppm)		(1000ppm)		
			Candida	Fuserium	Candida	Fuserium	
	albicans	oxysporium	albicans	oxysporium	albicans	oxysporium	
Ketokenazole	8mm	22mm	10mm	20mm	18mm	22mm	
6a	7mm	Nil	4mm	10mm	12mm	Nil	
6b	Nil	2mm	4mm	6mm	14mm	Nil	
6c	3mm	4mm	6mm	4mm	14mm	4mm	
6d	4mm	8mm	15mm	10mm	20mm	6mm	
6e	Nil	Nil	4mm	12mm	30mm	Nil	
6f	2mm	4mm	Nil	10mm	20mm	Nil	
6g	2mm	Nil	Nil	Nil	Nil	6mm	
6h	Nil	Nil	Nil	Nil	Nil	6mm	

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M. Mangal et al. / Heterocyclic Letters Vol. 6| No.4|795-803|Aug-Oct| 2016

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